

Binary Compounds

- | | |
|-----------------------|----------------------|
| 1. Magnesium Fluoride | 2. Potassium Oxide |
| 3. Silver Bromide | 4. Calcium Chloride |
| 5. Cesium Nitride | 6. Lithium Iodide |
| 7. NaCl | 8. RaO |
| 9. CsF | 10. K ₂ O |
| 11. SrCl ₂ | 12. MgI ₂ |

Polyatomic Compounds

- | | |
|------------------------|--|
| 13. Ammonium Chloride | 14. Calcium Sulfate |
| 15. Potassium Chromate | 16. Zinc Carbonate |
| 17. Strontium Arsenate | 18. Boron Nitrate |
| 19. LiCN | 20. Mg(HSO ₄) ₂ |
| 21. CaCrO ₄ | 22. Be(C ₂ H ₃ O ₂) ₂ |
| 23. KMnO ₄ | 24. Na ₂ CO ₃ |

Roman Numerals

- | | |
|------------------------------------|-----------------------------|
| 25. Manganese (VII) Chloride | 26. Cobalt (II) Bromide |
| 27. Titanium (IV) Iodide | 28.. Mercury (I) Nitrite |
| 29. Niobium (V) Acetate | 30. Rhodium (III) Phosphate |
| 31. CuO | 32. SiCl ₄ |
| 33. CrF ₆ | 34. FeBr ₃ |
| 35. Co ₃ N ₂ | 36. NiBr ₃ |

Prefixes

- | | |
|-----------------------------------|--------------------------------|
| 37. Carbon Monoxide | 38. Diphosphorous Trioxide |
| 39. Nitrogen Difluoride | 40. Diphosphorous Pentasulfide |
| 41. Dinitrogen trioxide | 42. Nitrogen Pentabromide |
| 43. N ₂ O ₅ | 44. PCl ₅ |
| 45. CO ₂ | 46. NO |
| 47. CCl ₄ | 48. SO ₂ |